

"Pushing the Limits of RF Superconductivity"

Workshop ~ Argonne National Laboratory ~ 22-24 September 2004





Bernard VISENTIN

dap <u>nia</u> CEC saclay	OUTLINE	
<b>Pushing the Limits of RF Superconductivity</b> Workshop ~ Argonne Nat. Lab. ~ 22-24 Sept. 2004	Baking on Low, Medium and High Field Q-Slopes HF Chemistry on Low and High Field Q-Slopes Diffusion ( $C_S$ , $T$ , $t$ ) at High Field Q-Slope Integrated System for Cavity Baking	

### INTRODUCTION

Improve the cavity performances (higher Q, higher field, quench) Understand the  $Q_0(E_{acc})$  curve



Experiments on Nb cavities

Experiment on Nb cavities Theory Analysis on Nb samples

Frame for existing theories

Suggestion for new theories

Argument mainly based on Saclay experiments



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already published

Reference [n] new results





### BAKING AT LOW TEMPERATURE

Brief Description ~ Some Features to Keep in Mind

**1.** Soft Heat Treatment ( $110 / 120 \circ C - 50 / 60$  hours)

2. First discovered on cavity with standard chemistry (BCP)

 $(50\% R_{BCS} decrease - slight R_{res} increase)$ 



The phenomenon exists also on electropolished (EP) cavities

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# **B**AKING AND Q-SLOPES

Baking has an effect on the three Q-slopes (BCP and EP cavities):

\* enhancement at low field \* slight flattening at medium field \* strong improvement at high field



Without Baking : Electropolished Cavity Can Not Reach 40 MV/m







## Q-SLOPE AT HIGH FIELD

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### **DIFFUSION PROCESS**

Baking treatment :  $R_{BCS}$  (  $\ell$  ) decreases

decrease linked to baking time

 $\rightarrow$  diffusion



SRF Workshop 1999 [7] ( P. Kneisel )

#### The Element ? Oxygen is good candidate

- $Nb_2O_5$ , NbO, sub-oxides, O interstitials at the interface with Nb
- diffusion at low temperature (Palmer's Thesis 1988) and more recently on sample analysis by XPS
- diffusion depth @ 120°C / 60 h compatible with RF layer (50 nm, 2K)

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### **DIFFUSION PROCESS** (cont.)





Diffusion Equation (Fick's 2<sup>nd</sup> law)

Diffusion Coefficient (Arrhenius equation)

Where : C(x,t) is the concentration, E<sub>a</sub> the activation energy, R the universal gas constant, T the temperature.

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Analytic solution in few cases, specially for a semi-infinite solid : with the initial condition (bulk)  $C(x,0)=C_0$ 

and the boundary condition (surface)  $C(0,t)=C_s$ 

 $\frac{C-C_s}{C_0-C_s} = erf \frac{x}{2\sqrt{Dt}}$ 

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# DIFFUSION AND Q-SLOPE

#### First Element :

E<sub>acc</sub> onset for Q-Slope is linked to the diffusion parameters

$$C = C_{S} \operatorname{erfc} \frac{x}{2\sqrt{Dt}}$$
$$D = D_{0} e^{-E_{a}/RT}$$
$$\left\{ \begin{array}{c} C_{S}, T, t \end{array} \right\}$$





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# **DIFFUSION** AND Q-SLOPE (cont.)

#### Other Elements :

Change diffusion parameters

time  $\downarrow \qquad 60 \rightarrow 3 \text{ hours}$ T  $\uparrow \qquad \left\{ \begin{array}{c} 110 \rightarrow 145 \text{ °C} \\ 120 \rightarrow 160 \text{ °C} \end{array} \right.$ 

Baking ( atmospheric air )

Q-Slope change but 160 °C / 3 h is too much

Adjustment : 145 °C / 3h







### CONCLUSIONS

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#### **1 - HF chemical treatment on baked cavities :**

Clusters and I.T.E. theories are probably not involved to explain the Low and High Field Q-slope modifications by baking.

#### 2 - High field Q-slope and diffusion parameters :

Diffusion process as the explanation for the Q-slope improvement?

#### **3 – Integrated Baking to improve the process**

Baking ( air - 3 hours ) during the cavity preparation in clean room

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ACKNOWLEDGEMENTS

Special thanks to my Saclay colleagues for their technical support :

Alain Aspart – Yves Gasser – Jean Pierre Poupeau (Chemistry)

> Jean Pierre Charrier – Bernard Coadou (RF tests and Vacuum)





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### Q-SLOPE AT MEDIUM FIELD

Theory : Thermal dependence of  $R_S$ 

Q-Slope

Surface Treatment by NP or NS Chemistry (1:10) @  $110^{\circ}C / 1$  hour + HF (45')

**Enhancement of the medium Q-slope** 

(not well analyzed at this moment : surface pollution)

**Differences less marked between M and HF Q-Slopes** 



 $6 \text{ Nb} + 10 \text{ HNO}_3 \rightarrow 3 \text{ Nb}_2\text{O}_5 + 10 \text{ NO} + 5 \text{ H}_2\text{O}$ 

 $Nb_2O_5 + 2 H_3PO_4 \rightarrow Nb_2O_2(PO_4)_2 + 3 H_2O$ 

$$Nb_2O_5 + n H_2SO_4 \rightarrow Nb_2O_{5-n}(SO_4)_2 + n H_2O_{5-n}(SO_4)_2$$